

Physics 137A Section 1: Problem Set #1
Due: 5PM Friday Jan 31 in 2nd floor LeConte-Birge Cross-Over

Suggested Reading for this Week: This week's homework is on the historical origins of Quantum Mechanics. Griffiths does not cover this subject in his book.

- I have put on our web page an excellent review that was published in the SLAC beamline in Fall 2000. Please read *The Origins of the Quantum Theory* by Cathryn Carson from that review.
- Michael Fowler from University of Virginia has posted a good review of Black Body Radiation on his course web page. I have put a pointer to it on our web page.
- Chapter 1 of Bransden and Joachain is very complete and goes into more details than I have in class. I have taken some of the problems for this week from there.

Homework Problems:

1. (Bransden Problem 1.6) In Bransden and Joachain and in the Fowler article, the equation that describes the “spectral distribution,” $\rho(\lambda, T)$, the energy per unit volume released in electromagnetic radiation in the wavelength interval between λ and $\lambda + d\lambda$ for a blackbody of temperature T is:

$$\rho(\lambda, T) = \frac{8\pi hc}{\lambda^5} \frac{1}{\exp(hc/\lambda kT) - 1}$$

where h is Planck's constant, c is the speed of light and k is the Boltzman constant. Prove that the total energy density $\rho_{TOT} = aT^4$ where $a = 8\pi^5 k^4 / 15h^3 c^3$.

Hint: $\int_0^\infty \frac{x^3}{e^x - 1} dx = \frac{\pi^4}{15}$

2. (Bransden Problem 1.13) In a photoelectric experiment in which monochromatic light of wavelength λ falls on a potassium surface, it is seen that it requires a potential of 1.91 V to stop an electron that is produced if $\lambda = 300$ nm is used. A potential of 0.88 V is required if $\lambda = 400$ nm.

- (a) Find a value for Planck's constant. You may use the fact that the charge of an electron is 1.6×10^{-19} C.
 - (b) Find the work function for potassium (the work function is the energy required to pull an electron from the surface of the metal)
 - (c) What is the threshold frequency for potassium (ie the lowest frequency for which it is possible to pull an electron from the metal)
3. Show that a free electron cannot absorb a photon and conserve both energy and momentum in the process (hence the photoelectric effect requires a bound electron).
4. (Bransden Problem 1.15) An x-ray photon of wavelength $\lambda_0 = 1$ Angstrom is incident on a free electron which is initially at rest. The photon is scattered at an angle $\theta = 30^\circ$ from the incident direction.
- (a) Calculate the Compton shift $\Delta\lambda$
 - (b) What is the angle ϕ (measured from the incident photon direction) at which the electron recoils?
 - (c) What is the kinetic energy of the recoiling electron?
 - (d) What fraction of its original energy does the photon lose?
5. (Bransden Problem 1.25) Calculate the de Broglie wavelength of:
- (a) A mass of 1 g moving with a velocity of 1 m/s
 - (b) A free electron having a kinetic energy of 200 eV
 - (c) A free electron having a kinetic energy of 50 GeV (1 GeV = 10^9 eV)
 - (d) A free proton having a kinetic energy of 200 eV

Hint: for (c) you must use the relativistic expression for momentum